Syllabus for Bachelor of Technology



Chemical Engineering

Subject Code: 01CH1402

Subject Name: Material Science and Engineering

B.Tech. Year – II (Semester IV)

Objective: Course will cover various aspects related to material, their properties and manufacturing methods used in chemical industries.

Credits Earned: 3 Credits

Course Outcomes: After completion of this course, student will be able to:

- 1. Understand the basic concept of crystal system as well as defects
- 2. Analyze the various properties of the engineering materials.
- 3. Relate the practical importance and relevance of metals and alloys in chemical industry.
- 4. Utilize the technological methods related to material strength and diffusion concepts.

Pre-requisite of course: None

Teaching and Examination Scheme

Teaching Scheme (Hours)				Theory Marks			Tutorial/ Practical		
							Marks		Total
			Credits	ESE	IA	CSE	Viva	Term work	Marks
Theory	Tutorial	Practical		(E)	(I)	(C)	(V)	(TW)	
3	0	0	3	50	30	20	0	0	100

Contents:

Unit	Topics		
		Hours	
1	Structure of Crystalline Solids		
	Amorphous & Crystalline solids, Unit cell, Crystal system and structure,		
	Coordination number, atomic packing factor, Point coordinates,	10	
	Crystallographic directions, Crystallographic planes-Miller indices.		
	Imperfections in solids: Defects and types of defects.		
2	Properties of Engineering Materials		
	Mechanical Properties: Isotropy, Anisotropy, Plasticity, Elasticity,		
	Toughness, Resilience, Tensile Strength, Malleability, Ductility, Brittleness,		
	Hardness, Fatigue, Creep, Mechanism of creep, wear resistance.		
	Electrical Properties: Resistivity, Conductivity (ionic & electrical),	6	
	Semiconductors, Superconductivity, Insulators, Ferro electricity, Energy		
	Bands in solids, classification of materials based on band gap.		
	Magnetic Properties: Magnetic moment, its origin, Ferro and Ferri-		
	magnetism, dia- and para-magnetism, ferrites, use of magnetic materials.		
3	Phase Diagram and Processing of Metal-Alloys	10	
	Phase diagram: Binary phase diagram, phase diagram for iron-carbon system,	10	

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	UNIVERSITY Chemical Engineering					
	Chemical Engineering					
	Development of microstructure in iron-carbon alloys.					
	Processing of Metal-Alloys: Ferrous alloys and nonferrous alloys, Fabrication					
	of metals, Thermal processing of metals, Corrosion in metal.					
4	Diffusion mechanisms					
	Diffusion mechanism, steady state diffusion, non-steady state diffusion,					
	Factors that influence diffusion, Diffusion in ionic and polymeric materials.	6				
	Rubber, Lubricants and adhesives					
	Definition, Characteristics, Types					
5	Ceramics and Composites					
	Ceramics: Ceramic structures, Properties of ceramics, Types and applications					
	of ceramics, Fabrication and processing of ceramics.	6				
	Composite Materials: particle-reinforced composites, fiber-reinforced					
	composites, structural composites, advantages and applications.					
	Total Hours	38				

References:

- 1. "Callister's Material Science and Engineering", 2/e R. Balasubramaniam, Wiley India.
- 2. "Elements of Material Science and Engineering", 6/e, Lawrence H. Van Vlack, Pearson Education.
- 3. "The Science and Engineering of Materials", 6/e, Donald R. Askeland and Pradeep P. Phule, Cengage Learning.
- 4. "Principles of Materials Science and Engineering", W F Smith, McGraw Hill.

Suggested Theory distribution:

M Marwadi

The suggested theory distribution as per Bloom's taxonomy is as follows. This distribution serves as guidelines for teachers and students to achieve effective teaching-learning process

Distribution of Theory for course delivery and evaluation							
Remember	Understand	Apply	Analyze	Evaluate	Create		
20%	35%	25%	20%	-	-		

Instructional Method:

- a. The course delivery method will depend upon the requirement of content and need of students. The teacher in addition to conventional teaching method by black board, may also use any of tools such as demonstration, role play, Quiz, brainstorming, MOOCs etc.
- b. The internal evaluation will be done on the basis of continuous evaluation of students in the laboratory and class-room.
- c. Practical examination will be conducted at the end of semester for evaluation of performance of students in laboratory.
- d. Students will use supplementary resources such as online videos, NPTEL videos, e-courses, Virtual Laboratory



Chemical Engineering

Design Based Problems (DP)/ Open Ended project (OEP) :

In the beginning of the session, subject faculty will allot an OEP / DP to the students. Students will be free to choose a topic of their choice which will be relevant to the syllabus and they will either prepare a working model/ report / presentation / poster on their topic.

Web Resources:

- a. https://nptel.ac.in/courses/113102080
- b. https://nptel.ac.in/courses/112108150